

9. Crystallography Symmetry

9.4. Splitting of atomic orbitals

Ref. Dresselhaus. Chap 5

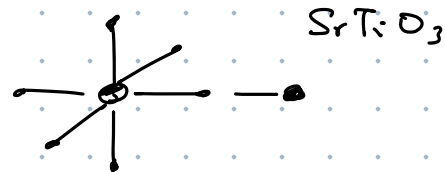
Ballhausen. intro. to ligand field theory Chap 4

Cowan. Theory of atomic structure and spectra

Consider an ion inside a crystal

$$H = H_f + V_{\text{crystal}}$$

↑
free ion



$$\textcircled{1} \quad H_f = \sum_i \left(\underbrace{\frac{p_i^2}{2m}}_{\text{kinetic}} - \underbrace{\frac{Z^+ e^2}{r_i}}_{\text{e-ion Coulomb}} + \sum_j \frac{e^2}{r_{ij}} + \underbrace{\sum_i \vec{l}_i \cdot \vec{s}_i}_{\text{spin-orbit}} \right) \quad \left(\begin{array}{l} \text{ignore} \\ \text{hyperfine} \\ \text{etc.} \end{array} \right)$$

with spherical symmetry. We know the solution.

② V_{crystal} : external potential due to other

ions in the crystal (Madelung potential)
"Crystal field" for ionic systems.

a. $V_{\text{crystal}} \propto \vec{l} \cdot \vec{s}$ rare-earths

b. $\sum \vec{l} \cdot \vec{s} \ll V$

transition metal compounds
(Cu, Ni, etc.)

magnetism, superconductivity

We focus on the second case. ignore SOC.

In this case, the symmetry-adapted basis functions are a better starting point than the atomic orbitals in spherical harmonics.

9.4.1. Splitting of the V^l rep.

For symmetry analysis, we can ignore the radial part of the basis functions, and consider spherical harmonics $Y_{lm}(\vartheta, \phi)$

$$Y_{lm} = \left[\frac{2l+1}{4\pi} \frac{(l-m)!}{(l+m)!} \right]^{\frac{1}{2}} P_l^m(\cos\vartheta) e^{im\phi}$$

$$\left(\int_0^{2\pi} d\phi \int_0^\pi \sin\vartheta d\vartheta Y_{lm} \overline{Y_{l'm'}} = \delta_{ll'} \delta_{mm'} \right)$$

each l labels an irrep

$$\hat{D}_R Y_{l,m} = \sum D^l(R)_{m'm} Y_{l,m'} \quad \text{Wigner D-matrix}$$

① rotation around \hat{z} by α then

$$\hat{D}_R Y_{l,m}(\vartheta, \phi) = Y_{l,m}(R^{-1}(\vartheta, \phi)) = e^{-im\alpha} Y_{l,m}(\vartheta, \phi)$$

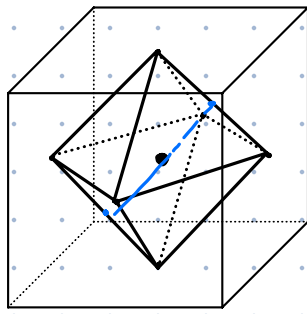
(we've seen it before)

$$X^l(\alpha) = \sum_m e^{-im\alpha} = \sum_{m=0}^l e^{-im\alpha} + \sum_{m=1}^l e^{im\alpha} = \frac{z^{l+1} - z^{-l-1}}{z - z^{-1}} = \frac{\sin(l+\frac{1}{2})\alpha}{\sin\frac{1}{2}\alpha}$$

$= U_l(\cos\frac{\alpha}{2})$
chebyshev polynomial
of the 2nd kind

② inversion $\hat{O}_i \cdot Y_{lm}(\vartheta, \phi) = Y_{lm}(\pi - \vartheta, \phi + \pi) = (-1)^l Y_{lm}(\vartheta, \phi)$

$O_h = m\bar{3}m$
 $4/m\bar{3}2/m$



C_3^{\pm} 3 body diagonals $\times \pm = 6$

C_2 connecting edge midpoints $\times 6$

$C_4(\pm)$ $3 \times C_4^{\pm} = 6$

$C_2(\pm)$ 3

$24 \times f.e.i = 48$

Now consider O_h with the character table below
 $C_2(\pm)$

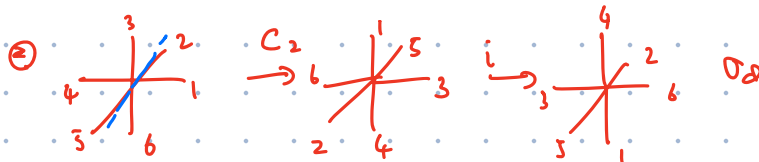
O_h	E	$8C_3$	$6C_2$	$6C_4$	$3C_2=(C_4)^2$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$	linear functions, rotations	quadratic functions	cubic functions
A_{1g}	+1	+1	+1	+1	+1	+1	+1	+1	+1	+1	-	$x^2+y^2+z^2$	-
A_{2g}	+1	+1	-1	-1	+1	+1	-1	+1	+1	-1	-	-	-
E_g	+2	-1	0	0	+2	+2	0	-1	+2	0	-	$(2z^2-x^2-y^2, x^2-y^2)$	-
T_{1g}	+3	0	-1	+1	-1	+3	+1	0	-1	-1	(R_x, R_y, R_z)	-	-
T_{2g}	+3	0	+1	-1	-1	+3	-1	0	-1	+1	-	(xz, yz, xy)	-
A_{1u}	+1	+1	+1	+1	+1	-1	-1	-1	-1	-1	-	-	-
A_{2u}	+1	+1	-1	-1	+1	-1	+1	-1	-1	+1	-	-	xyz
E_u	+2	-1	0	0	+2	-2	0	+1	-2	0	-	-	-
T_{1u}	+3	0	-1	+1	-1	-3	-1	0	+1	+1	(x, y, z)	-	$(x^3, y^3, z^3) [x(z^2+y^2), y(z^2+x^2), z(x^2+y^2)]$
T_{2u}	+3	0	+1	-1	-1	-3	+1	0	+1	-1	-	-	$[x(z^2-y^2), y(z^2-x^2), z(x^2-y^2)]$

$\left(\frac{\sin(l+\frac{1}{2})\alpha}{\sin\frac{1}{2}\alpha} \right)$

$\dim(2l+1) \quad \frac{2}{3}\pi \quad \pi \quad \frac{\pi}{2} \quad \pi$
 $\chi(E) \quad \chi(C_3) \quad \chi(C_2) \quad \chi(C_4) \quad \chi(C_2)$

$i \quad iC_3 = S_6 \quad iC_2(\pm) = \sigma_d \quad iC_4 = S_4 \quad iC_2(z) = \sigma_h$

① $iC_n = \sigma C_2 C_n = \sigma C_{2n}^{n+2} = S_{2n}^{n+2}$ odd; $iC_n = \sigma C_2 C_n = \sigma C_n^{\frac{n}{2}+1} = S_n^{\frac{n}{2}+1}$



	$\dim(2l+1)$	$\frac{2}{3}\pi$	π	$\frac{2}{2}\pi$	π	
	$\chi(E)$	$\chi(C_3)$	$\chi(C_2)$	$\chi(C_4)$	$\chi(C_2)$	
	i	S_6	σ_d	S_4	σ_h	
① s-orbital $l=0$	1 1	1 1	1 1	1 1	1 1	$s \rightarrow A_{1g}$
② p/ $l=1$	3 -3	0 0	-1 1	1 -1	-1 1	$p \rightarrow T_{1u}$
③ d/ $l=2$	5	-1	1	-1	1	$d \rightarrow E_g + T_{2g}$
						$E_g: x^2-y^2, 3z^2-r^2$
						$T_{2g}: xy, yz, xz$
④ f/ $l=3$	7	1	-1	-1	-1	$f \rightarrow A_{2u} + T_{1u} + T_{2u}$

9.4.2 Single d-electron in Octahedral field.

$$l=2. \quad Y_l^m = \sqrt{\frac{5}{4\pi} \frac{(2-m)!}{(2+m)!}} P_l^m(\cos\theta) e^{im\phi}$$

ignore

$$P_2^2(z) = 3(1-z^2)$$

$$P_2^{-m}(z) = (-1)^m \frac{(2-m)!}{(2+m)!} P_2^m$$

$$P_2^1(z) = -3z(1-z^2)^{1/2}$$

$$P_2^0(z) = \frac{1}{2}(3z^2-1)$$

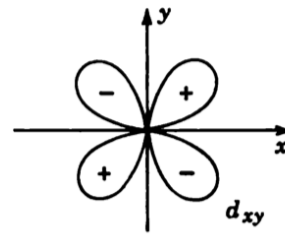
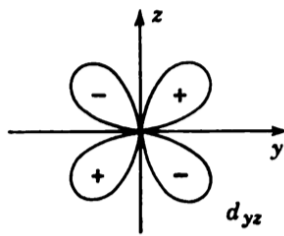
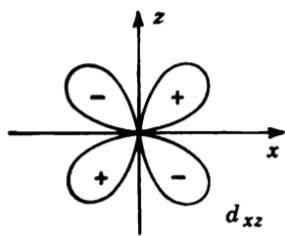
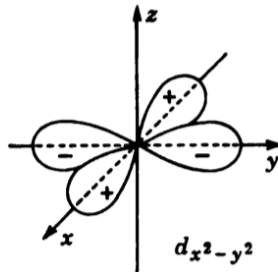
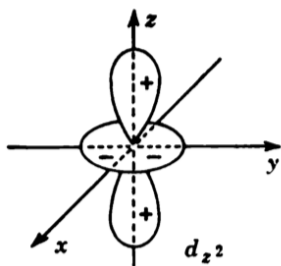
$$\left\{ \begin{aligned} d_{\pm 2} &= Y_2^{\pm 2} = \sqrt{\frac{3}{8}} (x \pm iy)^2 \\ d_{\pm 1} &= Y_2^{\pm 1} = \mp \sqrt{\frac{3}{2}} (x \pm iy)z \\ d_0 &= Y_2^0 = \frac{1}{2}(3z^2-1) \end{aligned} \right.$$

write in x,y,z.

$$\left\{ \begin{aligned} x &= \sin\theta \cos\phi \\ y &= \sin\theta \sin\phi \\ z &= \cos\theta \end{aligned} \right.$$

$$E_g = \begin{cases} x^2 - y^2 = \frac{1}{\sqrt{2}} (d_{+2} + d_{-2}) & \sim \sin^2 \theta (\cos^2 \phi - \sin^2 \phi) \\ z^2 \rightarrow 3z^2 - 1 = d_{02} & \sim \cos^2 \theta \end{cases}$$

$$t_{2g} = \begin{cases} xy = \frac{1}{\sqrt{2}i} (d_{+2} - d_{-2}) & \sim \sin^2 \theta (\sin \phi \cos \phi) \\ xz = -\frac{1}{\sqrt{2}} (d_{+1} - d_{-1}) \\ yz = -\frac{1}{\sqrt{2}i} (d_{+1} + d_{-1}) \end{cases}$$



$$E \uparrow \quad \begin{array}{l} d \\ \times 5 \end{array} \begin{cases} x^2 & e_g \\ x^3 & t_{2g} \end{cases} \quad \begin{array}{l} E = 6Dq \\ E = -4Dq \end{array} \quad \left. \begin{array}{l} \text{historically } 10Dq \\ \text{splitting} \end{array} \right\}$$

the sign and amplitude determined by the physical details. $E_{eg} > E_{t_{2g}}$ because of charge distribution

Formally, we can expand V_{crystal} onto spherical harmonics

$$V_c = \sum_i \sum_{lm} Y_l^m(\hat{r}_i) R_{nl}(r_i)$$

$$= V_A + V_0$$

\uparrow \leftarrow
 $l=0$ part octahedral part.

It should have all symmetries of H. transform as A_{1g} .

$$(V_c = \sum_{\mu} \lambda^{\mu} A_{\mu})$$

The potential for a d-electron:

$$\langle l, m_1 | V_c | l, m_2 \rangle \propto \begin{pmatrix} 2 & l & 2 \\ m_1 & m & -m_2 \end{pmatrix} \quad l \leq 4$$

We can further ignore odd orders because

d electrons are even. ($l=2$)

Remaining terms are Y_2^m and Y_4^m .

Y_2^m : $E_g \oplus T_{2g}$. no A_{1g} component

Y_4^m : $\underline{A_{1g} \oplus E_g \oplus T_{1g} \oplus T_{2g}}$

\Rightarrow find that $V_0 = Y_4^0 + \sqrt{\frac{5}{14}} (Y_4^4 + Y_4^{-4})$

(axis dependent.)

Check: $\hat{C}_4 \begin{pmatrix} x \\ y \\ z \end{pmatrix} = \begin{pmatrix} y \\ -x \\ z \end{pmatrix}$

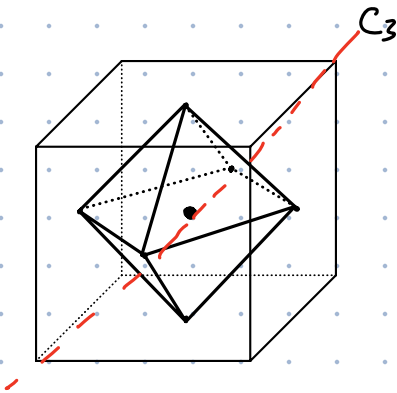
$$\hat{C}_4 Y_4^{\pm 6} = Y_4^{\pm 4} \quad Y_4^{\pm 4} \propto \sin^4 \theta e^{\pm i4\phi}$$

One can then write out the matrix elements of V_0 in d_m . ignore prefactors $\langle Y_2^m | Y_4^0 + \sqrt{\frac{3}{16}} (Y_4^4 + Y_4^{-4}) | Y_2^m \rangle$

$$\frac{1}{21} \begin{pmatrix} 1 & & & & 5 \\ & -4 & & & \\ & & 6 & & \\ 5 & & & -4 & \\ & & & & 1 \end{pmatrix} \quad E(E_g) = \frac{6}{21} = 6Dq$$

$$E(T_{2g}) = -\frac{4}{21} = -4Dq$$

Note that the form of E_g/T_{2g} orbitals depend on the quantization axis. if choosing C_3 as the



quantization axis. then

$$E_g = \begin{cases} \sqrt{\frac{1}{3}} d_2 + \sqrt{\frac{2}{3}} d_{-1} \\ \sqrt{\frac{1}{3}} d_{-2} - \sqrt{\frac{2}{3}} d_1 \end{cases}$$

$$T_{2g} = \begin{cases} d_0 \\ \sqrt{\frac{2}{3}} d_2 - \sqrt{\frac{1}{3}} d_1 \\ \sqrt{\frac{2}{3}} d_{-2} + \sqrt{\frac{1}{3}} d_1 \end{cases}$$

$$V_c = Y_4^0 + \sqrt{\frac{10}{7}} (Y_4^3 - Y_4^{-3})$$

$$(V_c)_{mm'} = \frac{1}{21} \begin{pmatrix} 1 & & & & \frac{5\sqrt{2}}{21} \\ & -4 & & & -\frac{5\sqrt{2}}{21} \\ & & 6 & & \\ \frac{5\sqrt{2}}{21} & & & -4 & \\ & & & & -\frac{5\sqrt{2}}{21} & 1 \end{pmatrix} \quad E(E_g) = -\frac{3}{7}$$

$$E(T_{2g}) = \frac{2}{7}$$

The sign of $6Dq$ is reversed.

Now consider $O_h \rightarrow D_{4h}$

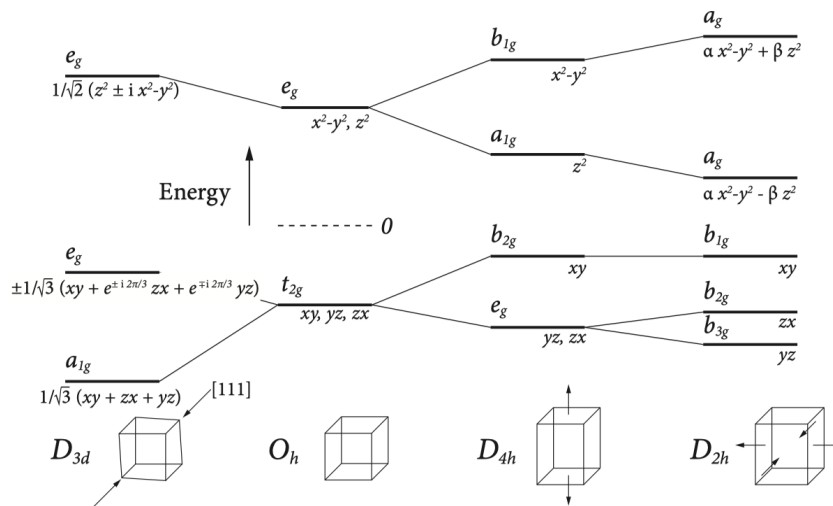


O_h	E	$8C_3$	$6C_2$	$6C_4$	$3C_2=(C_4)^2$	i
A_{1g}	+1	+1	+1	+1	+1	-
A_{2g}	+1	+1	-1	-1	+1	-
E_g	+2	-1	0	0	+2	-
T_{1g}	+3	0	-1	+1	-1	-
T_{2g}	+3	0	+1	-1	-1	-

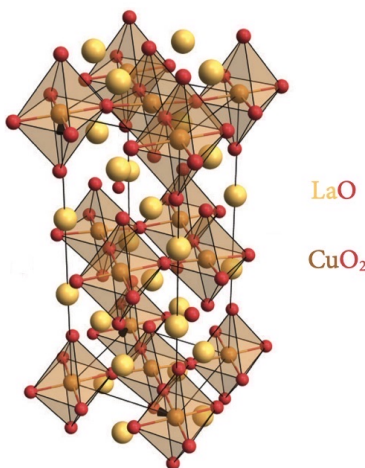
D_{4h}	E	$2C_4(z)$	C_2	$2C_2$	$2C''_2$	i	$2S_4$	σ_h	$2\sigma_v$	$2\sigma_d$	linear functions, rotations	quadratic functions	cubic functions
A_{1g}	+1	+1	+1	+1	+1	+1	+1	+1	+1	+1	-	x^2+y^2, z^2	-
A_{2g}	+1	+1	+1	-1	-1	+1	+1	+1	-1	-1	R_z	-	-
B_{1g}	+1	-1	+1	+1	-1	+1	-1	+1	+1	-1	-	x^2-y^2	-
B_{2g}	+1	-1	+1	-1	+1	+1	-1	+1	-1	+1	-	xy	-
E_g	+2	0	-2	0	0	+2	0	-2	0	0	(R_x, R_y)	(xz, yz)	-

$O_h \xrightarrow{T_{2g}} 2 \quad 0 \quad 2 \quad 2 \quad 0$
 $T_{2g} \xrightarrow{T_{2g}} 3 \quad 1 \quad -1 \quad -1 \quad -1$

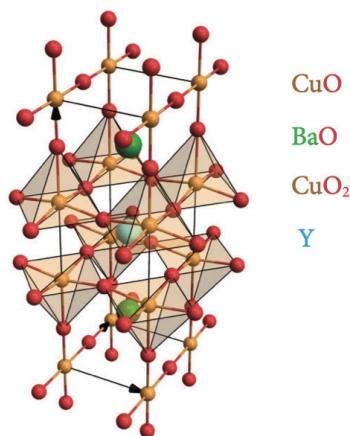
$O_h \quad D_{4h}$
 $E_g \rightarrow A_{1g} + B_{1g} \quad T_{2g} \rightarrow B_{2g} + E_g$
 $z^2 \quad x^2-y^2 \quad xy \quad xz, yz$



(a)



(b)



$Cu^{2+}: 3d^9$

$b_{2g} \uparrow \quad x^2-y^2$
 $a_{1g} \downarrow \quad 3z^2-r^2$

$b_{2g} \uparrow \downarrow \quad xy$
 $e_g \uparrow \downarrow \downarrow \quad xz/yz$

Figure 2.8 | Structures of (a) La₂CuO₄ and (b) YBa₂Cu₃O₇.

Character table for point group O_h

O_h	E	$8C_3$	$6C_2$	$6C_4$	$3C_2=(C_4)^2$	i	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$	linear functions, rotations	quadratic functions	cubic functions
A_{1g}	+1	+1	+1	+1	+1	+1	+1	+1	+1	+1	-	$x^2+y^2+z^2$	-
A_{2g}	+1	+1	-1	-1	+1	+1	-1	+1	+1	-1	-	-	-
E_g	+2	-1	0	0	+2	+2	0	-1	+2	0	-	$(2z^2-x^2-y^2, x^2-y^2)$	-
T_{1g}	+3	0	-1	+1	-1	+3	+1	0	-1	-1	(R_x, R_y, R_z)	-	-
T_{2g}	+3	0	+1	-1	-1	+3	-1	0	-1	+1	-	(xz, yz, xy)	-
A_{1u}	+1	+1	+1	+1	+1	-1	-1	-1	-1	-1	-	-	-
A_{2u}	+1	+1	-1	-1	+1	-1	+1	-1	-1	+1	-	-	xyz
E_u	+2	-1	0	0	+2	-2	0	+1	-2	0	-	-	-
T_{1u}	+3	0	-1	+1	-1	-3	-1	0	+1	+1	(x, y, z)	-	$(x^3, y^3, z^3) [x(z^2+y^2), y(z^2+x^2), z(x^2+y^2)]$
T_{2u}	+3	0	+1	-1	-1	-3	+1	0	+1	-1	-	-	$[x(z^2-y^2), y(z^2-x^2), z(x^2-y^2)]$

Character table for point group D_{4h}

(x axis coincident with C'_2 axis)

D_{4h}	E	$2C_4(z)$	C_2	$2C'_2$	$2C''_2$	i	$2S_4$	σ_h	$2\sigma_v$	$2\sigma_d$	linear functions, rotations	quadratic functions	cubic functions
A_{1g}	+1	+1	+1	+1	+1	+1	+1	+1	+1	+1	-	x^2+y^2, z^2	-
A_{2g}	+1	+1	+1	-1	-1	+1	+1	+1	-1	-1	R_z	-	-
B_{1g}	+1	-1	+1	+1	-1	+1	-1	+1	+1	-1	-	x^2-y^2	-
B_{2g}	+1	-1	+1	-1	+1	+1	-1	+1	-1	+1	-	xy	-
E_g	+2	0	-2	0	0	+2	0	-2	0	0	(R_x, R_y)	(xz, yz)	-
A_{1u}	+1	+1	+1	+1	+1	-1	-1	-1	-1	-1	-	-	-
A_{2u}	+1	+1	+1	-1	-1	-1	-1	-1	+1	+1	z	-	$z^3, z(x^2+y^2)$
B_{1u}	+1	-1	+1	+1	-1	-1	+1	-1	-1	+1	-	-	xyz
B_{2u}	+1	-1	+1	-1	+1	-1	+1	-1	+1	-1	-	-	$z(x^2-y^2)$
E_u	+2	0	-2	0	0	-2	0	+2	0	0	(x, y)	-	$(xz^2, yz^2) (xy^2, x^2y), (x^3, y^3)$

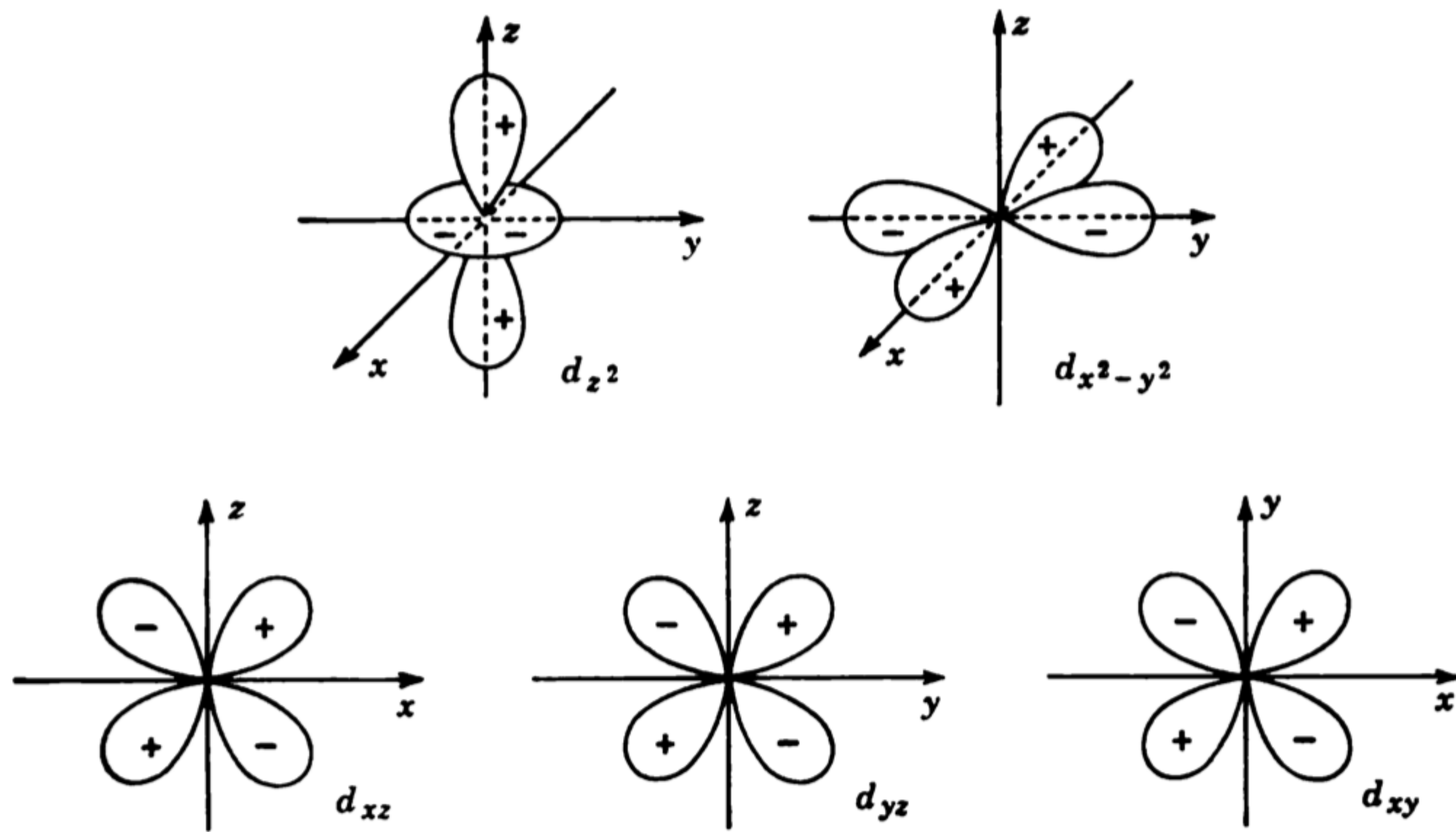
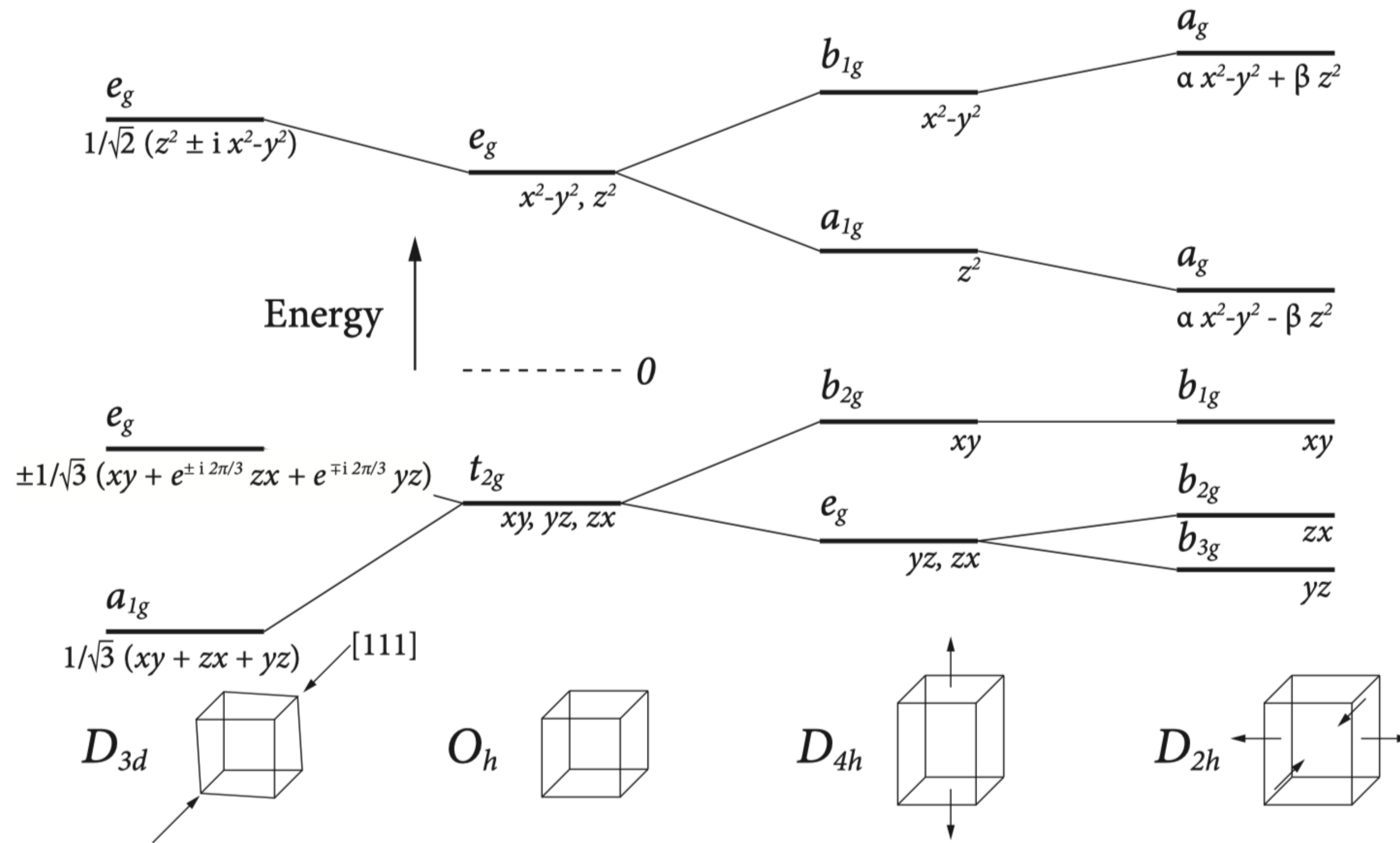
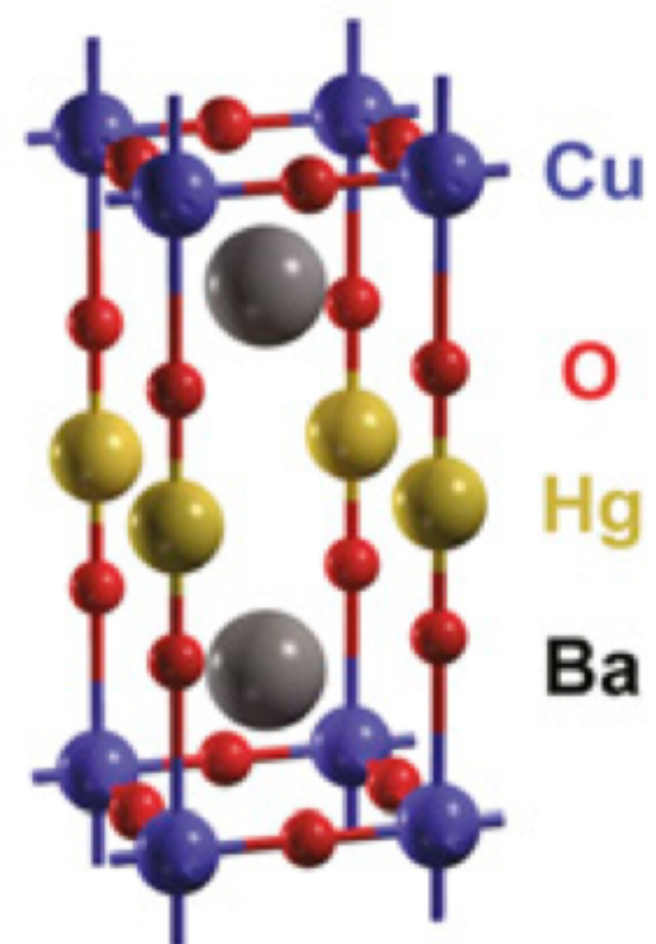


FIG. 4-1. e_g and t_{2g} electronic densities.

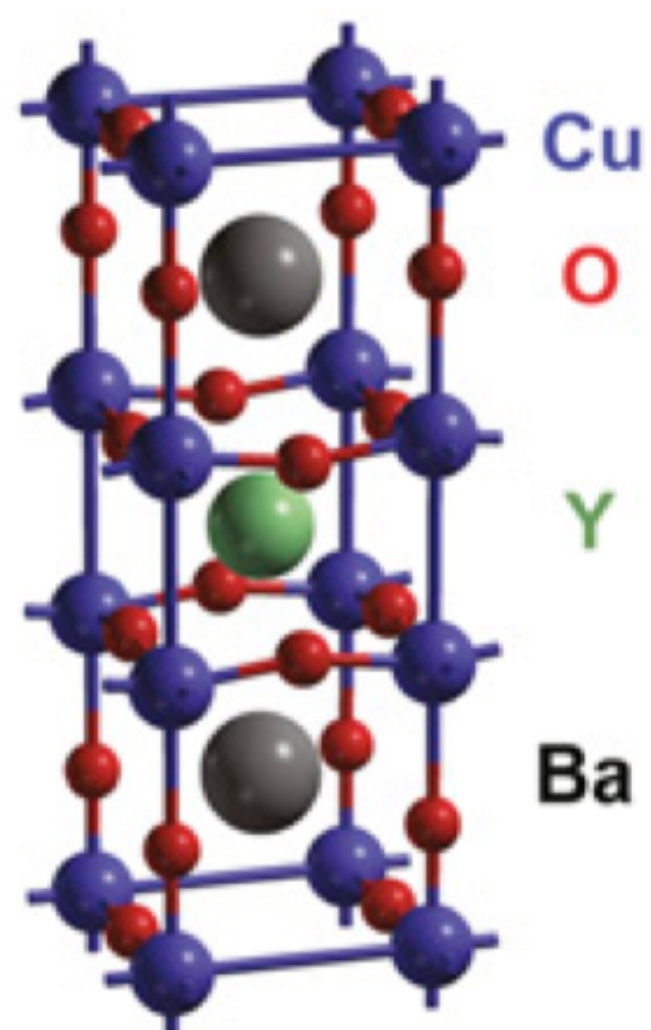


A

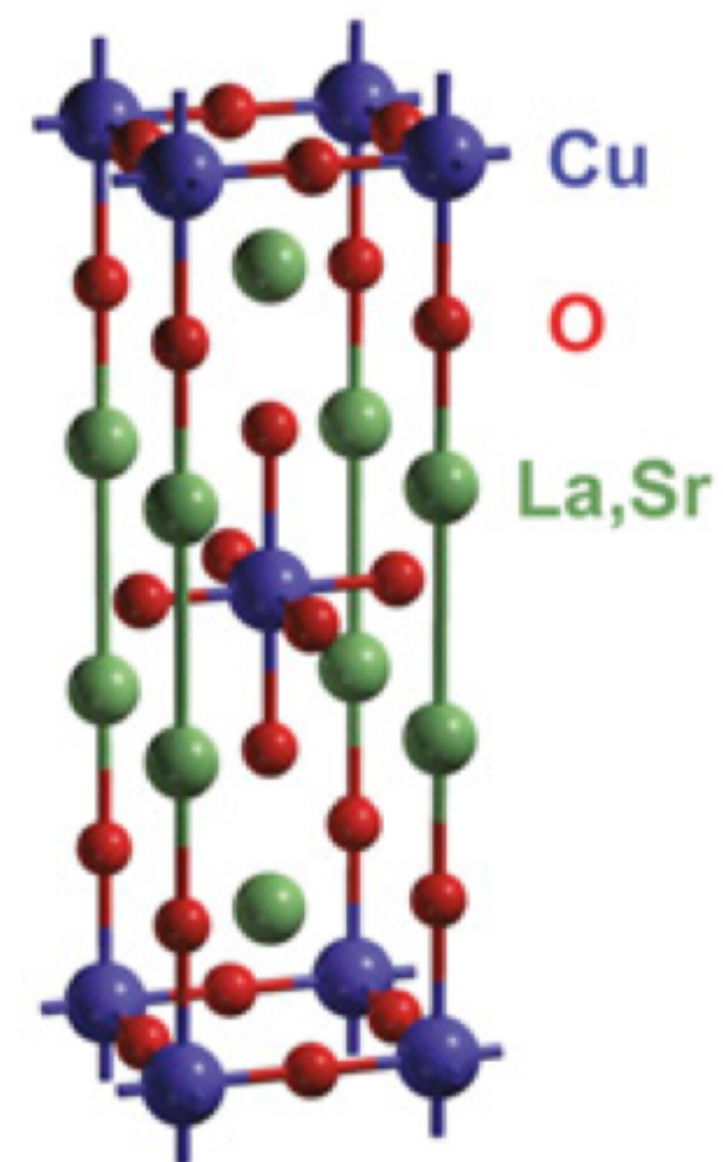
$\text{HgBa}_2\text{CuO}_{4+\delta}$
(Hg1201)



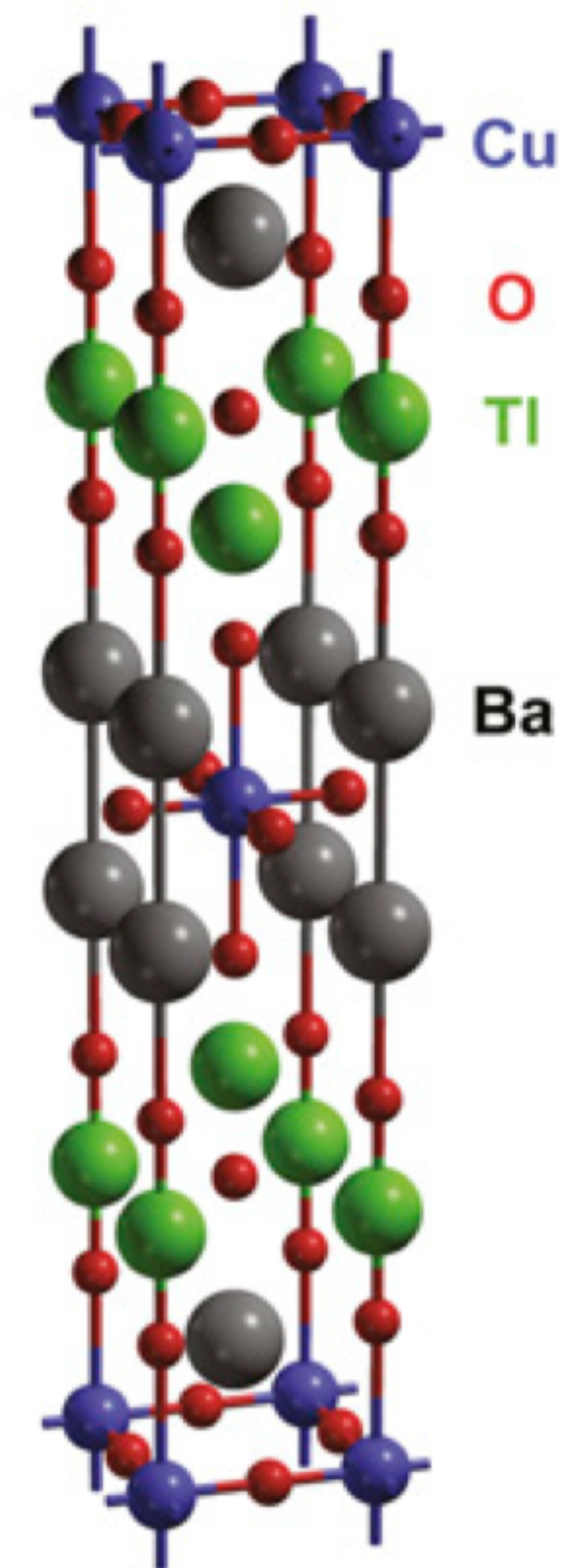
$\text{YBa}_2\text{Cu}_3\text{O}_{6+\delta}$
(YBCO)



$\text{La}_{2-x}\text{Sr}_x\text{CuO}_4$
(LSCO)



$\text{Tl}_2\text{Ba}_2\text{CuO}_{6+\delta}$
(Tl2201)



B

